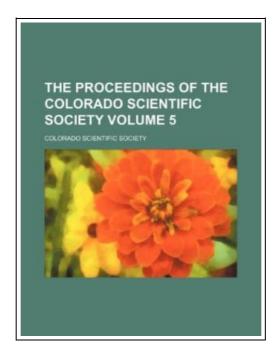
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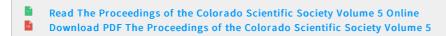
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Rarebooksclub.com, United States, 2012. Paperback. Book Condition: New. 246 x 189 mm. Language: English . Brand New Book ***** Print on Demand *****. This historic book may have numerous typos and missing text. Purchasers can download a free scanned copy of the original book (without typos) from the publisher. Not indexed. Not illustrated. 1896 Excerpt: . nor ignore its results. We are sometimes inclined to sneer at the mere analytical chemist; but let us remember that the analyst is always conservative, orthodox and safe; his work is the bed-rock of all chemical foundation, and without this safe foundation there can be no safe superstructure. It was the analytical work of such men as Bergman, Klaproth, Vauquelin, Wenzel, Richter and Proust, which prepared the chemical world for the atomic theory of Dalton. It was the analytical work of Gay-Lussac and Humboldt, on the volume-combination of gases, which laid the basis for the hypothesis of Avogadro and Ampere, that equal volumes of all gases, temperature and pressure being equal, contain the same number of molecules. It was the work of Berzelius in collating his own analytical results and that of others, which made him an authority on matters of chemical law and theory. In detail, it was the analysis of the mineral polluxite from Elba, in 1846, which makes Plattner the real discoverer of caesium: his figures footed up a total of only 92 75, yet he would not cook his figures although at the time they seemed inexplicable; we now know that be counted his caesium, (which was not identified at that time,) as sodium. Many other illustrations might be given, but we will not pause longer than to say that all modern chemistry is essentially quantitative; and hence, in a sense, all the various branches of organic, physiological, inorganic,...



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